

## Total Synthesis of ( $\pm$ )-Fenchone, a Synthesis of (+)-Fenchone and of (+)-*cis*-2,2,5-Trimethyl-3-vinylcyclopentanone, a Photoisomer of ( $-$ )-*trans*-Caran-4-one

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**Summary** (+)-*cis*-3-(2-Hydroxyethyl)-2,2,5-trimethylcyclopentanone and its racemic form are synthesised from (+)- $\alpha$ -2,3-epoxypinane or ( $-$ )-*trans*-pinocarveol, and ( $\pm$ )-2,2,5-trimethyl-1-oxocyclopentane-3-carboxylic acid, respectively; treatment with base of the chloroketone, derived from the (+)-hydroxyketone, affords (+)-fenchone, while pyrolysis of the (+)-acetoxycetone gives (+)-*cis*-2,2,5-trimethyl-3-vinylcyclopentanone, the latter and its *trans*-isomer being photo-products of ( $-$ )-*trans*- and ( $-$ )-*cis*-caran-4-one, respectively.

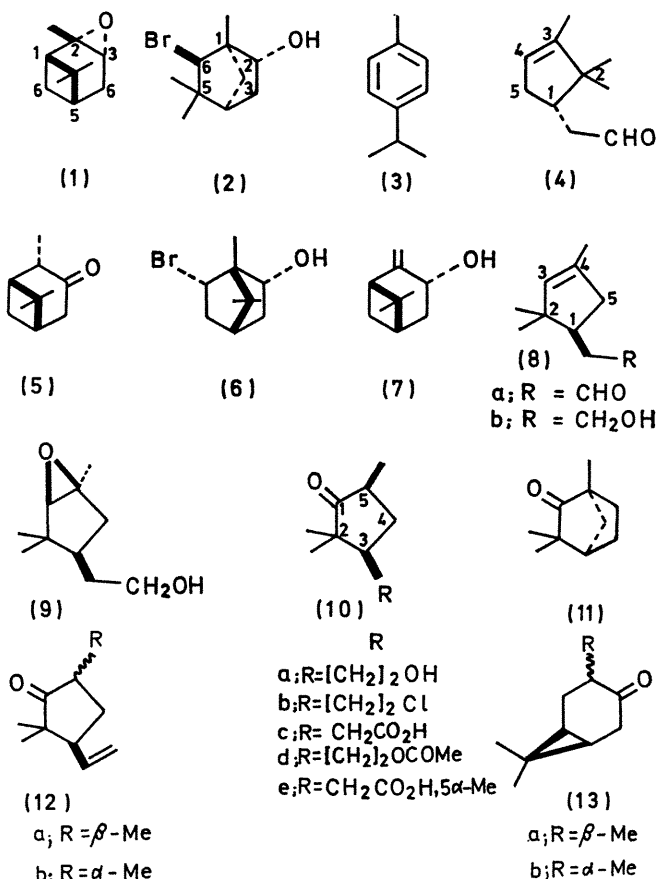
TREATMENT of (+)- $\alpha$ -2,3-epoxypinane (1) with ethereal hydrogen bromide yields ( $-$ )-6-*endo*-bromo-1,5,5-trimethylbicyclo[2,2,1]heptan-2-*exo*-ol (2) as major product, and *p*-cymene (3), ( $-$ )-2,2,3-trimethylcyclopent-3-enylacetaldehyde (4), (+)-pinocampone (5) and ( $-$ )-6-*endo*-bromo-1,7,7-trimethylbicyclo[2,2,1]heptan-2-*endo*-ol (6) in smaller yields. Similar treatment of ( $-$ )-*trans*-pinocarveol (7) gives slightly different proportions of the same products (cf. ref. 1).

Reaction of the bromo-alcohol (2) with silver acetate in acetic acid affords (cf. ref. 1) ( $-$ )-2,2,4-trimethylcyclopent-3-enylacetaldehyde (8a). Reduction of this with lithium aluminium hydride gives the alcohol (8b) which with peroxyacetic acid gives an almost quantitative yield of the ( $-$ )-*cis*-epoxy-alcohol (9) (cf. ref. 2), slightly contaminated with its acetate. Assignment of configuration of (9) is based on the intramolecular hydrogen bonding manifested in its i.r. spectrum in  $\text{CCl}_4$ . Isomerisation of (9) with  $\text{BF}_3 \cdot \text{Et}_2\text{O}$  gives (+)-*cis*-3-(2-hydroxyethyl)-2,2,5-trimethylcyclopentanone (10a) in 70% yield, which with thionyl chloride and pyridine affords the chloro-ketone (10b). Treatment of the last with ethoxide gives (+)-fenchone (11).

Racemic *cis*-3-(2-hydroxyethyl)-2,2,5-trimethylcyclopentanone, identical in spectra with its (+)-form (10a), is formed from ( $\pm$ )-2,2,5-trimethyl-1-oxocyclopentane-3-carboxylic acid<sup>3</sup> by Arndt-Eistert chain extension to the corresponding acetic ester, alkaline hydrolysis of which gives mainly the racemic *cis*-acetic acid corresponding to the active form (10c). Diborane reduction of the mainly *cis*-racemic acid gives solely the *cis*-racemic alcohol. This completes a new total synthesis of racemic fenchone.

Pyrolysis at 480° of the (+)-acetate (10d), of the keto-alcohol (10a), gives (+)-*cis*-2,2,5-trimethyl-3-vinylcyclopentanone (12a) as principal product (> 80% by g.l.c.). A

recent report<sup>4</sup> claims the synthesis of a similar cyclopentanone of unspecified configuration.



The corresponding *trans*-ketone (12b) was previously shown<sup>5,6</sup> to be the major photo-product of ( $-$ )-*cis*-caran-4-one (13a), its configuration being correctly assigned on mechanistic grounds.<sup>6</sup> It is, however, not stable to base as we earlier believed,<sup>5</sup> for we have converted it with sodium methoxide in methanol into the *cis*-isomer (12a), which is also obtained by the photolysis of ( $-$ )-*trans*-caran-4-one (13b).

We assign the configurations of the *cis*- and *trans*-ketones (**12a**) and (**12b**) (a) on mechanistic grounds (*cf.* ref. 6), (b) by analogy with the *cis*- and *trans*-2,2,5-trimethyl-1-oxocyclopentane-3-acetic acids (**10c** and **e**), the *cis*-form being the more stable,<sup>7</sup> and (c) on o.r.d. evidence, the molecular amplitude of the Cotton effect of the *cis*-ketone (**12a**) (+ 66·8) being greater than that of the *trans*-ketone (**12b**)

(+ 50·8), as is the case with the corresponding oxocyclopentaneacetic acids (**10c**) (+ 96) and (**10e**) (+ 27) respectively.<sup>8</sup>

Satisfactory spectra were obtained for the compounds described.

(Received, March 1st, 1971; Com. 167.)

<sup>1</sup> M. P. Hartshorn and A. F. A. Wallis, *J. Chem. Soc.*, 1964, 5254.

<sup>2</sup> H. B. Henbest, *Proc. Chem. Soc.*, 1963, 159.

<sup>3</sup> P. Hirsjarvi, *Ann. Acad. Sci. Fennicae*, 1952, A 11, 7.

<sup>4</sup> D. J. Goldsmith and R. C. Joines, *J. Org. Chem.*, 1970, 35, 3572.

<sup>5</sup> M. S. Carson, W. Cocker, S. M. Evans, and P. V. R. Shannon, *Tetrahedron Letters*, 1968, 6153.

<sup>6</sup> D. C. Heckert and P. J. Kropp, *J. Amer. Chem. Soc.*, 1968, 90, 4911.

<sup>7</sup> M. Harispe, D. Mea, A. Horeau, and J. Jacques, *Bull. Soc. chim. France*, 1963, 472.

<sup>8</sup> C. Ouannes and J. Jacques, *Bull. Soc. chim. France*, 1965, 3611.